## Fluids & Electrolytes

Richard J. Kagan, M.D.

Professor of Surgery
University of Cincinnati College of Medicine

# Body Water (as % body weight)

	Infant	Male	Female
Thin	80	65	55
Avg.	70	60	50
Obese	65	55	45

### **Functional Fluid Compartments**

EXAMPLE: 70 Kg man, TBW = 42 L (60% x TBW)

	Body Wt	Compartment	Volume
	5%	Plasma	3500 ml
ECF	15%	ISF	10500 ml
ICF	40%	ICF	28000 ml

Blood Volume = 7-8 % Body wt

Hct = 
$$\frac{RBC \text{ volume}}{RBC \text{ volume} + Plasma \text{ volume}}$$

## Replacement of a 500 ml Plasma Volume Deficit

Solution	Amount
LR	2000 ml
FFP	500 ml
5% albumin	500 ml
25% albumin	100 ml

## **Osmolarity**

Principal force of fluid movement

- Depends on # of active particles in solution that cannot pass through the semipermeable cell membrane
- Normal serum value = 285-300 mOsm/L
  - approximation: 2 (Na) + BUN/2.8 + Glucose/18
- Urine: 70-1200 mOsm/L
  - primarily controlled by ADH

# Water Balance

#### Intake

Oral: 800 - 1500 ml as liquids

500 - 700 ml as solids

Water of oxidation: 200 - 400 ml/day

## Water Balance

#### Output

Urine: 1000 - 2500 ml/day

⇒ need 500 - 800 ml to excrete products of catabolism

Water of stool: 100 - 200 ml/day

Insensible: 600 - 900 ml/day 25% = Respiratory (250 - 750 ml/day) 75% = Skin (400 - 600 ml/day)

# Causes of Increased Insensible Water Losses

- Fever: < 250 ml per °F above normal</li>
- Excessive evaporative skin losses (burns)
- Operation: 500 ml/hr
- Respirator or tracheostomy (up to 1.5 L/day)
- Hypermetabolism

# **Monitoring of Water Balance**

 Urine output: 0.5-1.0 ml/kg/hr (30-50 ml/hr - adults)

- Daily weights
- Hematocrit, electrolytes, osmolarity
- CVP, PCWP, Cardiac Output

## **Baseline Fluid Requirements**

CHILDREN: 1st 10 Kg → 100 ml/Kg/day

2nd 10 Kg → 50 ml/Kg/day

> 20 Kg → 20 ml/Kg/day

ADULTS: 30 - 35 ml/Kg/day

## **Baseline Electrolyte Requirements**

#### CHILDREN: < 20 Kg:

- Sodium: 2-3 mEq/Kg/day
- Potassium: 1-2 mEq/Kg/day
- Chloride: 1-1.5 mEq/Kg/day

#### **ADULTS:**

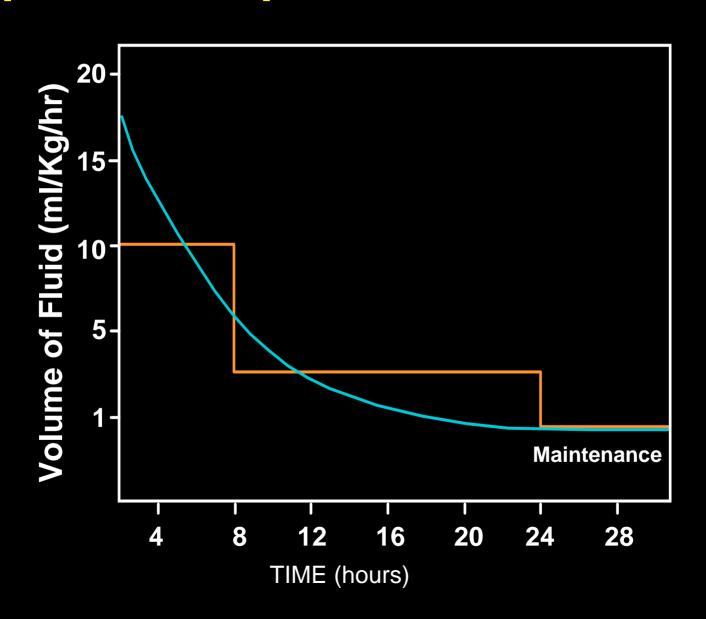
- Sodium: 75 120 mEq/day
- Potassium: 65 90 mEq/day
- Chloride: 85 145 mEq/day

## Other Body Fluid Losses

G.I. Tract: secretes 8-10 L/day

Source	Volume	Na	K	CI	$HCO_3$
Saliva	1500	10	25	15	30
Stomach	1500	60	10	130	
Duodenum	100-2000	140	5	80	
Pancreas	100-800	140	5	75	115
Bile	50-800	145	5	100	35
lleum	3000	140	5	104	30
Colon		60	30	40	

### **Graphic Concept of Fluid Resuscitation**



# Concept of 3<sup>rd</sup> Space

- > Fluid still within ECF (interstitial)
- Sequestered into areas of injury
- Must be replaced
- Losses are mobilized after recovery

## **Acute Dehydration**

# Most Common Fluid Disorder in Surgical Patients Diagnosis:

- 2% body wt: dry skin, thirst, 1 urine osmolarity, oliguria
- 4% body wt: dry tongue and axillae, oliguria, postural hypotension, tachycardia
- 6% body wt: LIFE THREATENING, above symptoms with lethargy, ileus

- Rapid infusion of balanced salt solution
- Avoid glucose-containing solutions

## **Chronic Dehydration**

May tolerate losses up to 10% of body wt.

More even distribution between ECF & ICF

Less symptomatic than acute

Treatment: Slowly over a period of days

# Hypotonic Dehydration (DESALTING WATER LOSS)

#### Most common fluid balance disorder

#### **Etiology:**

- Isotonic extrarenal losses
  - G.I. obstruction or fistula, ileus, vomiting, massive diarrhea, trauma
- Renal losses
  - osmotic diuresis, diuretic phase of ATN

# **Hypotonic Dehydration**

Diagnosis: hypotension, tachycardia, atony, obtundation, hypothermia; low serum Na, urine Na <20 mEq/L

- Prevention:
  - replace GI losses Q 4-6hrs if massive
  - approximate electrolyte content of fluid lost
- Therapeutic:
  - RAPID replacement of water and electrolytes

# Hypotonic Overhydration (Water intoxication)

#### **Etiology:**

- Overadministration of water
- Postop patients sensitive to hypotonic fluids
- Chronic visceral disease on low Na diet
- Chronic caloric starvation
  - increased endogenous water production
  - increased ADH
  - defect in Na pump due to energy deficit
- Inappropriate ADH syndrome
  - cancer, COPD, porphyria, head injury

## **Hypotonic Overhydration**

Diagnosis: lethargy, stupor, coma, convulsions, weight gain, anasarca, pulmonary edema

#### Treatment:

Water intoxication: Decrease water intake
 ? Solute diuretic (mannitol)
 ?? 3% NaCl, 1/6 M Na Lactate

 Inappropriate ADH: Decrease water intake Lithium Declomycin

# Hypertonic Dehydration (Desiccation)

#### **Etiology:**

- Evaporative water loss: respiratory tract fever, burn wound
- Loss of hypotonic fluids: excess sweating (50 mEq Na/L) diarrhea (in children)
- Renal abnormalities or abnormal renal stimuli (i.e., diabetes insipidus)
- Excess solute loading: hyperalimentation (NKHC, DKA) osmotic diuretics, angio. dyes
- Water deprivation

## **Hypertonic Dehydration**

#### Diagnosis:

- Dry mucus membranes, thirst, oliguria, CNS changes
- Increased BUN, Na, osmolarity

- Water replacement
- Monitor body weight, osmolarity, renal function
- Watch for signs of cerebral edema

## **Parenteral Solutions**

Solution	Na	K	Ca	Mg	CI	HCO <sub>3</sub>
ECF	142	4	5	3	103	27
LR	130	4	3	_	109	28
0.9% NaCl	154	-	-	-	154	
1/2 NS	77	-	-	-	77	
3% NaCl	513	-	_	-	513	

## Hyperkalemia

### Etiology:

- Renal insufficiency, stored blood
- Severe injury, cellular damage
- Metabolic acidosis (in exchange for H+)
- latrogenic

Symptoms: nausea, vomiting, diarrhea, colic weakness, depressed DTR's

## Hyperkalemia

Diagnosis: peaked T waves; prolonged ST, QRS

Complications: arrhythmias, ventricular fibrillation

- Glucose-insulin-bicarbonate: 50 cc D<sub>50</sub>
   10-25 units regular insulin
   1 amp NaHCO<sub>3</sub>
- Calcium gluconate (10%): 50-100 cc
- Kayexalate: 5-10 gm po QID or 30 gm enema Q 3-6h
- Dialysis: when K+ > 7 mEq/L

## Hypokalemia

#### **Etiology:**

- GI tract losses, alkalosis
- Renal excretion (diuretics)
- K into cells (insulin)
- Administration of K-free fluids
- Hyperaldosteronism
- Chronic hypercortisolism

Symptoms: nausea, vomiting, weakness

ileus, depressed DTR's

# Hypokalemia

Diagnosis: flat T wave, prolonged QT, U wave

**Complications: DIGITALIS TOXICITY** 

Treatment: potassium supplements (< 20 mEq/hr)

## **Hypercalcemia**

#### **Etiology:**

- Hyperparathyroidism
- Bony metastases
- Tumors secreting pseudoparathormone

Symptoms: CNS (weakness)
GI (anorexia)

- 0.9% Saline, furosemide
- Mithramycin
- Chelating agents
- Steroids

## Hypocalcemia

#### **Etiology:**

- Hypoparathyroidism
- Soft tissue infections
- Pancreatitis
- Renal failure

Symptoms: circumoral numbness, tingling hyperreflexia, tetany, cramps

Treatment: Calcium chloride or gluconate

## **Metabolic Acidosis**

#### **Etiology:**

- Increased organic acid
  - diabetic ketoacidosis, lactic acidosis
  - cellular hypoxia (shock, airway obstruction, sepsis)
- Decreased renal excretion of acid
  - oliguric renal failure
  - hyperchloremia
  - obstructive uropathy
- Abnormal loss of bicarbonate
  - diarrhea
  - small bowel or pancreatic fistula

#### **Metabolic Acidosis**

#### Compensation:

- Hyperventilation, early
- Renal, late

- Correct underlying problem
- NaHCO<sub>3</sub>
- Na lactate
- THAM (alkali w/o Na)

### **Metabolic Alkalosis**

#### **Etiology:**

- Loss of HCI
  - -NG suction, vomiting
- Loss of KCI
  - -vomiting, diarrhea
  - -diuretics, steroids, aldosteronism

### **Metabolic Alkalosis**

#### Compensation:

- generally uncompensated by lungs
- renal excretion of bicarbonate

- Chloride replacement
- Potassium replacement
- 0.1 N HCl or ammonium chloride

# Respiratory Acidosis

#### Etiology:

- Hypoventilation
- A-V shunting

Compensation: renal

Treatment: mechanical ventilation

# Respiratory Alkalosis

#### **Etiology:**

- pain

- fever

- sepsis

- early ARDS

Compensation: renal

- Pain management
- Ventilatory support (CPAP/PEEP)
- Search for possible underlying sepsis

# Interpretation of Blood Gases

	рН	pCO <sub>2</sub>	HCO <sub>3</sub>
Metabolic alkalosis	<b>†</b>		<b>†</b>
Metabolic acidosis	<b>↓</b>		<b>↓</b>
Respiratory acidosis	<b>↓</b>	<b>†</b>	
Respiratory alkalosis	<b>†</b>	•	

## **Analysis of Acid-Base Balance**

• pCO<sub>2</sub>  $\uparrow$  10 mm Hg > 40  $\Rightarrow$  pH decreases 0.08

•  $HCO_3 \uparrow 10 \text{ mEq} > 25 \Rightarrow \text{pH increases } 0.15$ 

- Bicarbonate deficit (mEq):
  - > (25 HCO<sub>3</sub>) x (0.2 x Body wt) or B.E. x (0.3 x Body wt)
  - ➤ Replace ≤ 50% at a time